

Pollution and Global Warming

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Degradation of methyl orange in water by visible excitation of a $\text{Bi}_2\text{WO}_6\text{-Fe}_2\text{O}_3$ catalystAsmae Bouziani¹, José Navio², Abderrahim El Hourch¹ and Mohammed El Azzouzi¹¹Mohammed V University, Morocco²University of Seville, Spain

The development of new photocatalysts is attracting vast interest. Among them the Bismuth tungstate (Bi_2WO_6) is a typical n-type direct band gap semiconductor with a band gap of 2.75 eV and has prospective applications in electrode materials, solar energy conversion and catalysis. In addition it has been found that Bi_2WO_6 might act the same as a stable photocatalyst for the photochemical decomposition of organic contaminants under visible light irradiation. The present study intended to dope Bi_2WO_6 with Fe_2O_3 to prepare $\text{Bi}_2\text{WO}_6\text{-Fe}_2\text{O}_3$ composite and investigate the degradation mechanisms of the composite in the presence of H_2O_2 . Methyl orange (MO) is used to imitate no biodegradable, toxic organic compounds. The photocatalytic activity of $\text{Bi}_2\text{WO}_6\text{-Fe}_2\text{O}_3$ to MO in the absence and presence of H_2O_2 is evaluated. $\text{Bi}_2\text{WO}_6\text{-Fe}_2\text{O}_3$ composites were synthesized using a mechanical mixing; by adding the Bi_2WO_6 obtained by hydrothermal method to the corresponding amount of Fe_2O_3 and their photocatalytic activity to degrade methyl orange (MO) under visible light illumination supported with H_2O_2 were studied. The H_2O_2 react with photogenerated electrons leading to the production of hydroxyl radicals ($\text{OH}\cdot$). The Fe_2O_3 acts like a Fenton reagent, accelerating the production of $\text{OH}\cdot$. $\text{Bi}_2\text{WO}_6\text{-Fe}_2\text{O}_3/\text{H}_2\text{O}_2$ system demonstrate much higher photocatalytic efficiency to degrade MO than pure Bi_2WO_6 . 50% of MO was degraded in 120 min visible irradiation and the peak disappeared after 90 min, and in $\text{Bi}_2\text{WO}_6\text{-Fe}_2\text{O}_3$ system the peak disappeared after 90 min. This study was performed at $\text{pH}\approx 6$ using Fe_2O_3 as a heterogeneous photo-Fenton catalyst in neutral and weak alkaline conditions of wastewater

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