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DC conductivity of Pb_xSe_{1-x}thin films prepared by thermal evaporation technique

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ABSTRACT

In the present work, including study the effect of different concentration (X=0.1, 0.2, 0.3, and 0.4), substrate temperature(303 and 348) K, and annealed to 348 0 K . on the electrical properties for $Pb_{x}Se_{1-x}$ thin films, which prepared by thermal evaporation. D.C. conductivity showed increase with increasing concentration, substrate temperature and annealing temperature and all films have two activation energy Hall measurements showed that the films at x=0.1 and x=0.2 was n- type, but at x=0.3 and x=0.4 was p- type with carrier's concentration and mobility increase with increasing of substrate temperature and annealing temperature and annealing temperature and annealing temperature and substrate temperature and su

Keywords; Lead chalcogenides, Optical absorption, Band gap, DC conductivity, lead salts, Barrier height

INTRODUCTION

Metal chalcogenide compounds, which are semiconductor in nature, are of considerable technical interest in the field of during the past decades[1]. The lead chalcogenides PbX (X=S, Se, Te) have been a subject of a great amount of theoretical and experimental studies, motivated by their importance in infrared technology, and more recently, because of their utility in laser technology and as thermoelectric materials [2,3].

The lead salts exhibit properties which are unusual, relative to other semiconductors. Compared for example with the usual III-V compounds, these IV-VI chalcogens present non typical electronic and transport properties, such as higher carrier motilities, higher dielectric constants, narrow band gaps and a positive temperature coefficients ,electronics and electro-optical devices [1,2,4].

Lead selenide is a polar semiconductor which has mixed ionic and overcomes the covalent bond where lead atoms connected to selenium atoms in dual ions form(Pb⁺² and Se⁻²)[3,5]. It is possible to prepare n-type and p-type of PbSe thin film. The PbX compounds are narrow direct gap semiconductors ,group IV-VI , which crystalline at ambient conditions in the cubic NaCl structure. The space lattice is face center cubic (f.c.c.) with lattice parameter(a= $6.122A^\circ$), energy gap equals 0.27eV at room temperature, dielectric constant is 20, the melting point of PbSe is 1338° K and the refractive index at (3µm) is 4.54 [6,7].

In this study the effect of concentration of selenium, substrate temperature and annealed temperature on the electrical properties of Pb_xSe_{1-x} thin films were investigated.

MATERIALS AND METHODS

The Pb_XSe_{1-X} with different X(0.1, 0.2, 0.3, 0.4) compound were prepared as alloys by using high purity (99,999%) (lead and selenium) metal obtained from Balzeres company. Each element weighted according to its atomic weight and then mixing in quartz tube (length=25 cm, diameter=0.9 cm) evacuated at pressure of (10⁻³ mbar). The tube was

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sealed and heated in electrical program controller furnace of type (Qallenhamp) at temperature 950 K and maintained at this temperature for about 5 hours and then allowed to cool slowly to room temperature . After that the ampoule was broken and the prepared compound of Pb_xSe_{1-x} was taken out and powdered to grain powder. This powder was used to prepare the films by thermal evaporation using Edward coating unit (model E306A) at a pressure of (10^{-5}) Torr. A molybdenum boat is used as the evaporation source and the substrates are placed directly above the source at a distance of nearly 15cm. The glass substrates is freshly cleaned with a pure alcohol and distilled water followed by ultrasonic agitation.

The electrical conductivity has been measured as a function of temperature for Pb_XSe_{1-X} films. The measurements have been done using sensitive digital electrometer type keithley (616) and electrical oven. The activation energies was calculated from the slope of the plot of $ln\sigma$ versus 1000/T according to equation [8].:

$$\sigma = \sigma_{\rm o} \exp\left(-E_{\rm a}/K_{\rm B}T\right) \tag{1}$$

Where σ_o is the minimum electrical conductivity at $0^0 K$, E_a is the activation energy, T is the temperature and K_B is the Boltzmann's constant .

Hall Effect measurement was determined by using HMS3000 Hall measurement setting. Applying a magnetic field `B' [called "Hall field] perpendicular to the electric field , yields a current (I) then the transverse electric voltage is given by [9].

$$R_{\rm H} = \frac{V_{\rm H}}{I} \cdot \frac{t}{B} \tag{2}$$

where I is Hall current, V_H is Hall voltage and R_H is Hall coefficient. From the Hall coefficient equation we can determine the carrier's concentration of the semiconductor, and the carrier type, since RH is negative for n-type and positive for p- type :

For n-type
$$R_{\rm H}$$
=-1/n.e (3)

For p-type
$$R_{\rm H}=1/p.e$$
 (4)

If the conduction is due to one carriers type e.g. electrons, then the mobility can measure [10]:

$$\mu_{n} = \frac{\sigma}{n.e} = \sigma \left| \mathbf{R}_{\mathrm{H}} \right| \tag{5}$$

RESULTS AND DISCUSSION

The crystal structure of Pb_XSe_{1-X} films, which were evaporated on glass substrates with thickness (150±20) nm at room temperature were shown in figure (1). The structure become a mixture of cubic and hexagonal at (X=0.2,0.3 and 0.4)). The preferred orientation lies along (200) direction. Our results are nearly in agreement with Al-Woaely[11] and Kassim etal [12].

The d.c conductivity for Pb_XSe_{1-X} films has been studied as a function of $10^3/T$ with the range of (300-473 K), at different value of X (0.1, 0.2, 0.3, 0.4), different substrate temperatures (303, and 348) K, and annealed to 348 ⁰K. as shown in figure(2). It is clear from these figure that there were two transport mechanisms, giving rise to two activation energies E_{a1} and E_{a2} . At higher temperature range (331–393) K, the conduction mechanism is due to carrier excited into the extended states beyond the mobility edge. At lower temperature range (291 – 331) K; the conduction mechanism is due to carrier excited into localized states at the edge of the band. We can review the results as follows: From figure the ($\sigma_{d.c}$) of Pb_XSe_{1-X} films increases as the Pb concentration (x) increases. This may be due to change in the localized states, structure, and composition of films as well as to the re-arrangement of atoms which yields fewer defects. This results are agreement with Kumar etal[13].



Fig. (1)X-ray diffraction patterns of Pb_xSe_{1x} prepared at room temperature with different x (0.1, 0.2, 0.3 and 0.4)



 $Fig.(2) \ ln \ \sigma \ versus \ 1000/T \ for \ for \ Pb_xSe_{1-x} \ (A) \ T_a \ \& \ T_S \ at \ RT \ , (B) \ T_a = 348k \ and \ T_S = RT(C), \ T_a = RT \ and \ T_S = 348K, \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \ at \ 348K \ (D) \ T_a \ \& \ T_S \$

It ass found that the $\sigma_{d.c}$ conductivity of Pb_XSe_{1-X} films with different x increases with increasing the substrate temperatures (T_s) and annealing temperature(T_a), The activation energies decreases with increasing the substrate temperature and annealing(as shown in figure 3) due to decreases the energy gap.



 $\begin{array}{l} Fig.(3)(a) \ \sigma_{d.c} \ vs. \ Pb \ concentration \ \&(\ b) \ E_{a1} \ and \ E_{a2} \ vs. \ Pb \ concentration \ for \ Pb_xSe_{1-x} \ at \ (A) \ T_a \ \& \ T_S \ at \ RT, \ (B) \ T_a=348 k \ and \ T_S=RT, \ (C) \ T_a=RT \ and \ T_S=348 k \ , \ (D) \ T_a \ \& \ T_S \ at \ 348 K \end{array}$

The value of $(\sigma d.c)$ and activation energies are listed in Table (1).

Ta (°k)	Ts (°k)	х	σ _{RT*} 10 ⁻²	E _{a1} (eV)	Range (K)	E _{a2} (eV)	Range (K)
		0.1	0.026	0.0731	303-373	0.1725	373-473
RT	RT	0.2	0.187	0.0537	303-373	0.1538	373-473
		0.3	0.281	0.0434	303-373	0.1532	373-473
		0.4	1.456	0.0322	303-373	0.1389	373-473
		0.1	0.037	0.0750	303-373	0.1833	373-473
348k	RT	0.2	0.265	0.0546	303-373	0.1606	373-473
		0.3	0.498	0.0580	303-373	0.1357	373-473
		0.4	3.268	0.0366	303-373	0.1320	373-473
		0.1	0.042	0.0713	303-373	0.1487	373-473
RT	348k	0.2	0.299	0.0628	303-373	0.1406	373-473
		0.3	0.736	0.0686	303-373	0.1531	373-473
		0.4	3.831	0.0453	303-373	0.1254	373-473
		0.1	0.026	0.0705	303-373	0.1528	373-473
348k	348k	0.2	0.174	0.0401	303-373	0.1228	373-473
		0.3	1.333	0.0353	303-373	0.1156	373-473
		0.4	5.797	0.0281	303-373	0.1190	373-473

 $Table \mbox{(1): D.C Electrical conductivity and activation energies at different values of (x) for \mbox{Pb}_x\mbox{Se}_{1\cdot x} \ \ with the Effect of substrate and annealing temperature}$

The sign of Hall coefficient (R_H) indicates that $Pb_{1-x}Se_x$ thin films behaves as a n-type semiconductor for(x=0.1 and x=0.2) and p-type for(x=0.3 and x=0.4).

The carrier's mobility was calculated from equation (5). From the figure(4 a and b) it was found that all the carriers concentration increases with substrate temperature and annealing temperature ,but mobility decrease with substrate temperature increases.



(4)(a) and (b) Variation of Hall mobility and carrier concentration as the versus X for Pb_xSe_{1-x} films at (A) $T_a=RT$ and $T_S=RT$ (B) $T_a=348$ kand $T_S=848$ (C) $T_a=848$ kand $T_S=348$ K

This may due to the re-crystallization process, which leads to the decrease of defects in the film during the film growth, and consequently a decrease of the carriers scattering at the defect. The increasing of re-crystallization leads to rising the potential barrier, for that reason the mobility increasing. Our results are agreement with Nayef 1[4] and Arivazhagan etal[15].

It can be observed from table (2) that the carrier's concentration and mobility increasing with increase of the Pb concentration. The substrate temperature and annealing temperature considered as an effective parameter on the carrier concentration and mobility.

$T_a(\mathbf{q})$	$T_{s}(\mathbf{q})$	x	$R_H(cm^3/c)$	$n*10^{15}(cm^{-3})$	$\sigma (\Omega.cm)^{-1}$	$\mu(cm^{2}v.sec)$	type
		0.1	4.40E+03	1.42	0.00026	1.1	n
RT	RT	0.2	2.30E+03	2.72	0.001869	4.3	n
		0.3	4.10E+03	1.52	0.002806	11.5	р
		0.4	4.70E+03	1.33	0.014556	68.4	р
		0.1	3.70E+03	1.69	0.000372	1.4	n
348	RT	0.2	1.47E+03	4.25	0.002653	3.9	n
		0.3	2.54E+03	2.46	0.004975	12.6	р
		0.4	1.52E+03	4.13	0.03268	49.5	р
		0.1	3.00E+03	2.08	0.000421	1.3	n
RT	348	0.2	4.40E+02	14.20	0.002988	1.3	n
		0.3	1.80E+02	34.72	0.007358	1.3	р
		0.4	2.30E+02	27.17	0.038314	8.8	р
		0.1	1.50E+03	4.17	0.000261	0.4	n
348	348	0.2	2.20E+02	28.41	0.001736	0.4	n
		0.3	9.00E+01	69.44	0.013333	1.2	р
		0.4	1.15E+02	54.35	0.057971	6.7	р

Table: (2): Hall parameters of Pb_xSe_{1.x} at X=0.1,0.2,0.3and0.4 with the effect of substrate temperature and annealing temperature

CONCLUSION

The Pb_xSe_{1-x} with differnt X(0.1, 0.2, 0.3, 0.4) compound were prepared successfully as alloys by using high purity (99,999%) lead and selenium metal. The D.C. conductivity for all films increases as the Pb concentration, substrate temperature and annealing temperature increase . There are two transport mechanisms of the charge carriers. In general the activation energies decrease with increasing Pb concentration, substrate temperature and annealing temperature showed that Pb_xSe_{1-x} films were n-type at x=0.1 and x=0.2, and p-type at x=0.3 and x=0.4. Carrier's concentration and mobility increase with increasing with the Pb concentration. While the carrier's concentration increases with substrate temperature and annealing temperature, but mobility decreases

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