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Microbial desulphurization of crude oil using Aspergillus flavus

¹Adegunlola, G. A., ¹Oloke, J. K., ¹Majolagbe, O. N., ¹Adebayo, E. A., ²Adegunlola C. O., ²Adewoyin A. G and ³Adegunlola F. O.

 ¹Department of Pure and Applied Biology, Microbiology Unit, Ladoke Akintola University of Technology, Ogbomoso, Oyo State, Nigeria
²Department of Science Laboratory Technology, Ladoke Akintola University of Technology, Ogbomoso, Oyo State, Nigeria
³Department of Microbiology, University of Ilorin, Ilorin, Kwara State, Nigeria

ABSTRACT

The ability of the immobilized spores of Aspergillus flavus cultured in increasing concentrations of sodium metabisulphite to remove sulphur from crude oil was investigated. When 50g of immobilized spores of A. flavus was added to 100ml of crude oil sample for one, two, three and seven days, the amount of sulphur removed were 27.2%, 45.2%, 90.4% and 91.7% respectively. When 50g of immobilized spores of A. flavus was introduced into 100ml of crude oil for seven days at 35°C, 40°C and 45°C, the amount of sulphur removed were 63.2%, 55.3% and 10.5% respectively. Lastly, when 10g, 50g and 100g of immobilized spores of A. flavus were added to 100ml of crude oil for seven days, the amount of sulphur removed were 49.6%, 94.7% and 53.9% respectively. Hence, A. flavus can be used to reduce sulphur contents in crude oil to a minimal level.

Keywords: Microbial desulphurization, immobilized spores, metabisulphite, Biodesulphurization.

INTRODUCTION

Crude oil is a complex mixture of a large number of organic compounds including sulphur compounds. Most products of crude oil contain sulphur, especially the low grade petroleum fuel. When they are burnt, they produce sulphur dioxide (Sarojini, 2005). The total sulphur content of crude oil varies from reservoir to reservoir. Sulphur compounds form the largest group of non-hydrocarbon compounds and are by far the most important and expensive for the refiner to deal with (Maurice, 2003). Sulphur in crude oil plays a critical role in atmospheric pollution as it occurs in the form of acidic gases such as sulphur dioxide which may be harmful to health (Ukoli, 2000; Sarojini, 2005).

Environmental pollution and acid rain caused by the release of sulphur dioxide on combustion of sulphur-containing fossil fuels has brought about regulations for production of low sulphur in fuels. The level of sulphur in fuels is regulated to reduce sulphur related air pollution. To meet these regulations, sulphur must be removed from fuels during refining process. Organic sulphur in crude oil is removed by refineries using hydrodesulphurization – HDS (Atlas, 1981; Grossman, 2005; Shifflet *et al.*, 2000).

The fact that Hydrodesulphurization can only remove the bulk of inorganic sulphur and simple organic sulphur (Furaya, *et al.*, 2001; Kabe *et al.*, 1992) but inadequate to produce low sulphur fuels due to the fact that the complex polycyclic sulphur compound present in petroleum and coal is not removed by the process (HDS) accounts for the introduction of microbial desulphurization. Microbial desulphurization is a process where sulphur content of crude

oil is exposed to microorganisms that can specifically break carbon sulphur bond, thereby releasing sulphur in a water soluble, organic form (Folsom *et al.*, 1999; Galapher *et al.*, 1993). Actions of microorganisms on crude oil component transform contaminants into non-toxic compounds (Leahy and Colwell, 1990). Various components of crude oil can be used as a source of carbon and energy by microorganisms

Rhodococcus sp. and *Arthrobacter sulphureus* are bacteria used in desulphurization of fossil fuel (Izumi *et al.*, 1994). They are isolated from oil contaminated sludge/soil. They are used for reducing the sulphur content of diesel samples, thereby indicating their potential for use as biocatalysts in desulphurization of fossil fuels (Labana *et al.*, 2005). Immobilization method is used to entrap the spores of fungi so that it can be used for biodesulphurization. Immobilization makes organisms remain in their aqueous suspension.

MATERIALS AND METHODS

The experiment was carried out in the Laboratory of Pure and Applied Biology, Ladoke Akintola University of Technology (LAUTECH), Ogbomoso, Oyo State, Nigeria. Crude oil was collected from an oil producing company (Shell Petroleum Company, Warri, Delta State, Nigeria). The material comprises oily wastes such as drill cutting, tank bottom sludge and pure crude oil.

Training of Organisms: The organism used for this research work was an isolate of *A. flavus*. The organism was originally isolated from an oil contaminated environment. To train *A. flavus*, minimal salt broth was prepared by weighing 2.0g NH_4NO_3 , 1.5g KH_2PO_4 , 0.5g KCl, 1.5% glucose, 0.05g yeast extracts into 1000ml distil water. Sodium metabisulphite ($Na_2S_2O_5$) was added at different concentrations of 1g/l, 3g/l, 5g/l, 10g/l and 15g/l so that it can act as surrogate of the sulphur substances in crude-oil. The broth was poured into bottles and sterilized in an autoclave.

The organism was cultured in the minimal salt broth with the increasing concentration of sodium metabisulphite being the source of energy and sulphur. The culture was incubated at room temperature for 72 hours. Growing spores of *A. flavus* were harvested and grinded to form a uniform suspension.

Spores Immobilization: The spores of *A. flavus* was immobilized by mixing 80g of the harvested spores in 1% (w/v) sterile alginate solution and then gelled into beads by dropping the suspension into a cold $15g/litre CaCl_2$ solution.

Procedure for Sulphur Removal

The immobilized spore of *A. flavus* was used to desulphurize crude oil under three conditions. These include different time duration, temperature and different concentration of immobilized spores. The desulphurized crude oil was subjected to ultra violet visible spectrophotometric analysis.

Quantification of Sulphur: Biodesulphurized crude oil sample (2ml) was weighed in a conical flask and added to 10ml of concentrated HCl contained in Kjedah digestion flask. Distilled water (20ml) was then added. The contents was well shaken to hydrolyse and then allowed to stay for 3 hours. The content was filtered with No.1 Whatman filter paper. The filtrate was kept for analysis. The filtrate (5ml) was poured into a test tube. Distilled water (15ml) and 2ml of conditioning reagent were then added. The test tube was covered and allowed to stand for few hours. A Spatula full of BaCl₂ was then added. The turbidity was read with ultra violet visible spectrophotometer.

RESULTS

Sodium metabisulphite "trained" *A. flavus* was used to desulphurize crude oil under three different conditions. These include different time durations, temperatures and concentrations of immobilized spores.

Table 1 shows the effect of different time duration on sulphur removal from crude oil using *A. flavus*. When 100ml of crude oil sample was treated with 50g of the immobilized spores of *A. flavus* for one and two days, the amount of sulphur removed were 27.2% and 45.2% respectively (Table 1). When the experiment was run for three and seven days, the amount of sulphur removed were 90.4% and 91.7% respectively (Table 1). These two values show no significant difference from each other but significantly different from the values obtained for control and time durations (pr < 0.0001).

Table 2 shows the effect of different temperatures on sulphur removal from crude oil using *A. flavus*. When 50g of immobilized spores of *A. flavus* was introduced into the crude oil for 3 days at 35°C and 40°C, the amount of sulphur removed were 63.2% and 55.3% respectively (Table 2). The difference between the two values is insignificant (pr <

0.0001). At 45°C, the sulphur level in the crude oil was reduced by 10.53% (Table 2). There is a significant difference between the two values obtained at 40°C and 45°C (Pr<0.0001). The values obtained from the three different temperatures are significantly different from the control value (Pr<0.0001).

Table 3 reveals the effect of different amount of immobilized spores of *A. flavus*_on sulphur removal. When the crude oil sample was treated with 10g and 50g of immobilized spores of A. flavus for three days, the amount of sulphur removed were 49.6% and 94.7% respectively (Table 3). The statistical analysis of the results shows a significant difference between the two values obtained (Pr<0.0005).

TABLE 1: Effect of different time d	uration on sulphur removal	l in crude oil using A. Flavus

Duration of Experiment (Day)	Weight of Spores (g)	Absorbance Value	% Sulphur removed
1	50	0.190	27.2
2	50	0.185	45.2
3	50	0.159	90.4
7	50	0.137	91.7

TABLE 2: Effect of different temperature on sulphur removal in crude oil using A. Flavus

Temperature (°C)	Weight of Spores (g)	Absorbance Value	% Sulphur removed
35	50	0.160	53.2
40	50	0.162	55.3
45	50	1.819	10.5

TABLE 3: Effect of different amount of immobilized spores on sulphur removal in crude oil using A. Flavus

Weight of Spores	Duration of Experiment (Day)	Absorbance Value	% Sulphur removed
10	3	0.172	49.6
50	3	0.130	94.7
100	3	0.165	53.9

DISCUSSION

In this research work, the spores of *Aspergillus flavus* was entrapped using sodium alginate. This was done to immobilize the spores of *A. flavus*. The immobilized spores of *A. flavus* was used for the biodesulphurization process. The entrapment of spores in alginate used in this work is preferable and better than other entrapment methods such as adoption used in binding *Escherichia coli* spores on to an ion exchange resin. This is due to the fact that the entrapment method operates under a mild condition and also because of the simplicity of the procedure used.

When 50g of immobilized spores of *A. flavus* was used to treat 100ml of crude oil sample for three days and seven days, the amount of sulphur removed were 90.4% and 91.7% respectively (Table 1). It took longer time compared to the time taken by *Mycobacterium phlei WU-F1*, an organism that degraded 0.81mM Dibenzothiophene (Larbode and Gibson, 1977) just within 90 minutes. Also another report by Chang *et al*, (1998) showed that diesel oil was desulphurized in 48 hours using *Norcadia sp. strain* CYKS2.

In this research work, another parameter determined is the ability of *A. flavus* to biodesulphurize crude oil at different temperatures. Previous work on biodesulphurization was achieved at temperature as high as 50°C using thermophilic bacteria such as *Mycobacterium phlei WU-F1* and *Mycobacterium phlei WU-0103* (Furaya *et al*, 2001). In this study, *A. flavus* was able to biodesulphurize crude oil at 35°C. When 50g of immobilizerd spores of *A. flavus* was used to treat 100ml of crude oil at 35°C, the amount of sulphur removed was 63.2% (Table 2). Oldfield *et al*, (1997) reported that at mild temperature of 35°C, elemental removal seems probable because at higher temperature, enzymatic activity and catalytic property of the organism can be disrupted. In this research work, when temperature was increased to 40°C and 45°C, *A. flavus* was able to reduce the sulphur level of the oil by 55.3% and 10.5% (Table

2) respectively. It can be deduced from this result that the higher the temperature, the lower the amount of sulphur removed.

Grossman *et al*, (2001) reported that desulphurization using *Rhodococcus sp. strain ECRD-1* reduced the sulphur content of the oil by 30%. In this research work, when 50g and 100g immobilized spores of *A.flavus* were used to treat 100ml of crude oil, the amounts of sulphur removed were 94.7% and 54.0% respectively (Table 3). These amounts of sulphur removed from crude oil are high compared to the total sulphur contents removed in the oil as reported by Grossman *et al*, (2001).

This research work demonstrates contribution to knowledge by revealing the ability of *A. flavus* as a biological agent for the removal of sulphur from crude oil. It revealed the use of immobilization technology in trapping microorganisms yet maintaining their catalytic properties without any cell washed out.

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REFERENCES

[1] Atlas RM, (1981). Microbiological Review, 45 (1). 180-206.

[2] Chang JH, Rhee SK, Chank YK, Chang HN (**1998**). Desulphurization of Diesel Oils by a newly Dibenzothiophene Degrading *Norcadia sp. Strain* CYKS2. 851-855.

[3] Furaya T, Kirimura K, Kino K, Usami S (2001). WU-F1. 129-133.

[4] Folsom MJ, Lee RC, Prince KK, Garrette GN (1999). Applied Environ. Microbiol. 65.

[5] Galapher JR, Olson ES and Starley DC. (1993). FEMS Microbio. Lett. 107:31-36

[6] Grossman MJ, (2001). Microbial Removal of Original Sulphur from Fuels – a review of Past and Present Approaches 345 – 359: In M.L. Deelli and R. Chianelli: Hydro-treating Technology for Pollution Control Catalysts, Cotalysis and Processes. Mareel Dekker New York.

[7] Izumi Y, Olishiro T, Ogeno H, Hine Y. (1994). Appl. Environ. Microbiol. 223-226.

[8] Kabe T, Ishiara A, Tajima H, (1992). Industrial Eng. Chem. 1577 – 1580.

[9] Labana S, Pandey G, and Jain, RK. (2005). Desulphurization of Dibenzothiophene and diesel Oils by *Bacterials* Vol. 40. 159.

[10] Larbode A, Gibson DT. (**1977**). *Appl.Environ. Microbiol.* 783 – 790.

[11] Leahy JG, and Colwell RR (**1990**). *Microbiological Reviews*. 54 (3): 305 – 315.

[12] Oldfield C, Wood NT, Gibert SC, Murray FD, and Faure FR (**1997**). Desulphurization of Benzothophene and Dibenzothophene by *Actinomycete* Organisms belonging to the Genus *Rhodococcus* and other Related Taxa, 74: 111 - 132.

[13] Sarojini RR (2005): Modern Biology. AFP African-FEP Publishers Ltd. 132-161.

[14] Shifflet J, Guerinik K and Cobrum (**2000**): Deep Desulphurization of Extensively Hydrosulphurized Middle Distillate Oil by Rhodococcus Strain ECRD-1. *Appl Environ. Microbiology*.