



Industrial and Medical Applications of Enzymes

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DESCRIPTION

Enzymes are biological catalysts that accelerate chemical reactions in living organisms. They are essential for metabolic processes, DNA replication, digestion, and many other vital functions. Without enzymes, most biochemical reactions would be too slow to sustain life. Enzymes are highly specific, meaning each enzyme typically acts on a particular substrate to catalyze a specific reaction. Understanding enzymes is fundamental to biology, medicine, and biotechnology. The study of enzymes dates back to the 19th century. In 1833, Anselme Payen and Jean-François Persoz discovered the first enzyme, amylase, which breaks down starch into sugar. Later, in 1877, Wilhelm Kühne coined the term “enzyme” (from the Greek words “en” meaning “within” and “zyme” meaning “yeast”). In the early 20th century, researchers like James Sumner and John Northrop demonstrated that enzymes are proteins, paving the way for modern enzymology. Enzymes are primarily composed of proteins, though some RNA molecules, known as ribozymes, also exhibit enzymatic activity. Enzymes have unique three-dimensional structures that determine their specificity and function. The active site of an enzyme binds to the substrate, forming an enzyme-substrate complex, which then undergoes a chemical transformation to produce the final product. Enzymes work by lowering the activation energy required for a reaction. The general mechanism of enzyme action involves the enzyme binds to its specific substrate at the active site. Temporary interactions between the enzyme and substrate occur, often inducing a conformational change in the enzyme. The chemical reaction takes place, converting the substrate into the product. The product is released, and the enzyme is free to catalyze another reaction cycle. Enzymes are classified based on the type of reaction they catalyze. The six major classes include catalyze oxidation reduction reactions. Transfer functional groups between molecules. Catalyze hydrolysis reactions. Break chemical bonds without hydrolysis. Catalyze the rearrangement of atoms

within a molecule. Join two molecules together using ATP. Several factors influence enzyme activity, including each enzyme has an optimal temperature. Substances that decrease enzyme activity. They can be competitive or non-competitive. Enzyme activity is tightly regulated in cells to maintain homeostasis. Regulation occurs through molecules bind to sites other than the active site, altering enzyme activity. The end product of a metabolic pathway inhibits an earlier step, preventing overproduction. Enzymes can be activated or deactivated by phosphorylation, methylation, or other modifications. Enzymes have widespread applications in various industries. Enzymes are used in drug development, disease diagnostics, and treatments. Enzymes aid in food processing, such as amylases in baking, proteases in cheese-making, and pectinases in fruit juice clarification. Enzymes are essential for genetic engineering, PCR, and industrial bioprocessing. Enzymes help in bioremediation, breaking down pollutants and waste materials. Enzymes play a pivotal role in molecular biology techniques, including cut DNA at specific sequences, essential for cloning. Used in PCR to amplify DNA sequences. Join DNA fragments during recombinant DNA technology. Enzymes are indispensable to life, facilitating countless biochemical reactions with remarkable specificity and efficiency. Their study has revolutionized medicine, industry, and biotechnology, leading to advancements in health, food production, and environmental sustainability. Ongoing research continues to uncover novel enzymes with potential applications, making enzymology a continuously evolving field with vast implications for the future.

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CONFLICT OF INTEREST

The author declares there is no conflict of interest.

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